Today, we will:

- Discuss Rayleigh flow [heat added to (or removed from) a duct] qualitatively
- Discuss Rayleigh flow quantitatively

Rayleigh flow

Introduction: Approximations and Assumptions:

- One-D flow (ignore boundary layers V approx. constant at any cross-section of the duct, i.e, at any x location; so, V = V(x) only Otherwe need numerical solution
- Ideal gas

Control volume analysis:

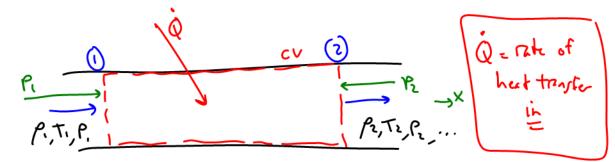
• Constant area duct (straight section of pipe)

• Constant gas properties $(\gamma, C_P, R, \text{etc.})$ even if chemical reactions or combustion provides the heat input (different gas properties of the combustion products and/or different mixture of gases after a reaction) - con overse the contacts after combusting

• Negligible friction along duct walls

is all other increasing the

(separation, turbulent dissipation, etc.)



Q >0 if heat allow Q<0 is heat removal

CONS. LAWS FOR THY

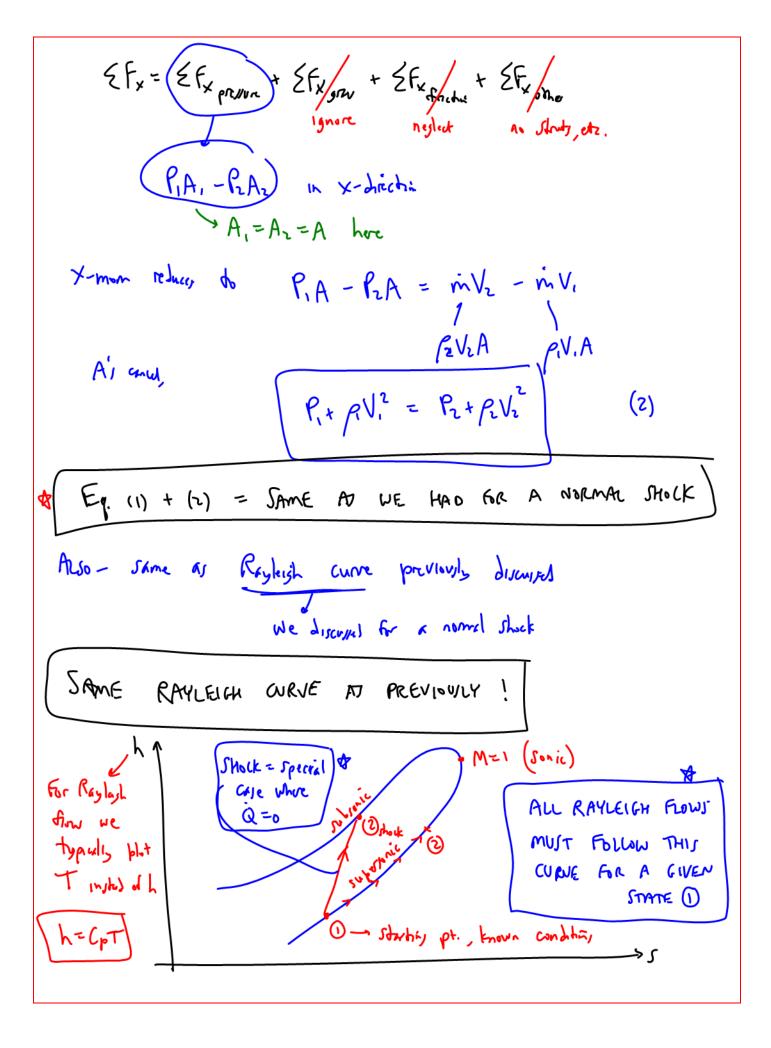
· may in, = in = content Since Azconst, PV= P2V2 (1)

EF = E pinV - E pinV

even if we inject
fuel i born to
generate Q we ignore
the allohimal in

(no licks)

ONE -0 OPPOX



· NORMAL SHOCK = ADIABATIC (Q=0) SUMMARY: : ho,=ho, => To,=Toz (Special case of Rylayd offin) THIS IS AT ONE SPECIAL POINT To shock ON OUR PLOT · RAYLEIGH FLOW (in general) Q = 0 : To, + To, \ ho, + h. " We follow the curve of we all or remove heat (d) Where 20 we end up on the ReyleyL arm (5654 (d)? @ Depas on how much Q ENERBY EQ $Q + \dot{m} \left(h_1 + \frac{V_1^2}{2} \right) = \dot{m} \left(h_2 + \frac{V_2^2}{2} \right)$ Let $g = \frac{\dot{Q}}{\dot{n}} \left[g = heat transfer per unit may of thus]$ "specific hut transfer (30) OR (Mu) h, + V12 = GpT, + V2

ALTERNATE (3) IS
$$Q = C_p(T_2-T_1) + \frac{V_2^2-V_1^2}{2}$$
 (36)

$$S_2 - S_i = G_p \ln \frac{T_2}{T_i} - R \ln \frac{P_2}{P_i}$$
(4)

Producilly we write
$$ds = \frac{\delta q}{T} + ds |mayor > 0$$

